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Total and Partial Absorption Coefficients for a Nitrogen Plasma

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TOTAL AND PARTIAL ABSORPTION COEFFICIENTS FOR A NITROGEN PLASMA

I. INTRODUCTION

When an intense electron beam is propagated through the atmosphere a hot, ionized channel is created. A detailed picture of the chemistry and physics within the channel is crucial to the understanding and modeling of electron beam propagation through the atmosphere.

An issue of current importance is channel cooling, specifically, to what extent does energy relax from the heated channel to the surrounding environment. To determine the role of radiation emission and transport through the channel, one must perform a detailed radiative transfer calculation. The primary ingredient for such a calculation is the absorption coefficient. For electron temperatures of ~ 3 e.v., for example, nitrogen and oxygen atoms and their ions are the dominant constituents of heated air. In this paper we present the atomic absorption coefficients for a nitrogen plasma using a model reported earlier [1]. The relevant photoabsorption and photoionization cross sections are also given.

Johnston et. al. [3] have studied the radiative properties of high temperature air in detail. Their calculations assume an equilibrium composition for air of varying densities and include thousands of line and photoionization transitions. While the emphasis of their

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work is for temperatures less than 1 e.v., and thus on accurately describing the molecular absorption coefficient, they extend the calculations to much higher temperatures where the atomic component is dominant. It is in this regime that we compare results.

II. ABSORPTION AND IONIZATION CROSS SECTIONS

Specific details of our model for the nitrogen plasma may be found elsewhere [1]. For electron temperatures of 1-3 e.v. we assume the plasma consists of N, N⁺, N⁺⁺, and electrons and include in our calculation the lowest thirteen (13) levels of N I, the lowest seventeen (17) levels of N II, and two representative levels for N III. This simple model for nitrogen includes the dominant uv and visible emissions arising from bound-bound and free-bound transitions. In this section the absorption cross sections for all bound-bound, bound-free, and free-free transitions contained within our model are described in detail.

A. Bound-Bound Transitions

The photoabsorption cross section for transitions between a lower level l and an upper level u is given [2] by,

$$\sigma_{vul} = \sigma_{vul,m} \frac{(\Delta v)^2}{(v - v_{ul})^2 + (\Delta v)^2} \quad (1)$$

where

$$\sigma_{vul,m} = \frac{\lambda^2}{8\pi^2} \frac{g_u}{g_l} \frac{A_{ul}}{\Delta v} \quad (2)$$

Equations (1) and (2) assume a Lorentzian line shape for the transition whose resonance frequency is ν_{ul} and wavelength is λ . The statistical weights of the upper and lower states are g_u and g_l , A_{ul} is the Einstein coefficient, and $\Delta\nu$ the line width. Contributions to line broadening include Stark and Doppler broadening. We calculate the contribution of each as a function of frequency, electron temperature, and electron density and choose $\Delta\nu$ to be the greater of the two. For example, at high temperatures and low electron densities, the Doppler width is greatest whereas larger electron densities cause the Stark effect to dominate. The Stark width is always larger for electron densities greater than $10^{18}/\text{cm}^3$. The necessary Stark widths were obtained from reference [4], either directly or through an interpolation procedure for those transitions included in our model but not listed in Griem's tables.

We recognize that in situations where there are two broadening mechanisms, one resulting in a Gaussian line shape and the other in a Lorentzian line shape, the proper description involves Voigt profiles with an appropriately defined width. However, the above model has been chosen for simplicity and is not expected to significantly alter the results.

Figures (1a) - (3a) show the bound-bound absorption cross sections for electron temperatures of 1.0, 2.0, and 3.0 e.v.. Details of the plasma composition (Ne, N, N⁺, N⁺⁺) are discussed in section III. For all three cases the lines are Stark broadened at all photon energies. As the electron temperature increases and the plasma contains more electrons the lines broaden with a corresponding decrease in the absorption maximum. At Te = 1.0 e.v. the cross sections peak at approximately 10⁻¹⁴ - 10⁻¹³ cm². For higher temperatures there is a drop of nearly two orders of magnitude.

B. Bound-Free Transitions

The photoionization cross sections have been obtained from several sources. The following functional form was introduced by Henry [5] and used by Ali [6] to describe photoionization from the ground state configurations of N,

$$\sigma_{\lambda} = \sigma_{th} [\alpha(\lambda/\lambda_0)^s + (1-\alpha)(\lambda/\lambda_0)^{s+1}] \times 10^{-18} \text{ cm}^2 \quad (3)$$

The parameters α , s , and σ_{th} can be found in reference [6b]. In our calculations, the threshold wavelengths λ_0 have been corrected for the reduction in ionization

potential due to the immersion of an isolated atom in a plasma in the manner described in reference [1].

The quantum-defect method [7] provides a good approximation to nonhydrogenic continuous absorption cross sections. Griem [7] has tabulated these cross sections for various atoms including neutral nitrogen. For bound-free transitions between $N^{Z-1}(n, \ell)$ and N^Z , where n and ℓ are the principal and angular momentum quantum numbers of the initial atomic or ionic bound state, the cross section is given by

$$\sigma_{n\ell}^{Z-1} = 15.8 \times 10^{-18} \frac{g^Z}{g_{n\ell}} \frac{n^3}{z^2} \frac{\nu_n^3}{\nu^3} f_{n\ell}(\nu) \quad (4)$$

Here, $g_{n\ell}$ is the statistical weight of the initial state, g^Z the statistical weight of the final state, ν the frequency of the photon, and ν_n the threshold transition frequency (again corrected for the ionization potential reduction). For neutral nitrogen, the frequency-dependent factor $f_{n\ell}(\nu)$, is computed using a linear interpolation procedure based on the information provided in reference [7]. For N^+ , an average value of 0.2 is used.

The results are shown in Figs. (1b) - (3b). Structurally, the cross sections are similar. Bound-free absorption is important for visible and uv frequencies. The onset of different absorptions depend on the threshold frequency; this in turn depends on the electron density and

temperature through the the ionization potential reduction. The cross section varies between 10^{-17} and 10^{-16} cm^2 , and is a small contribution compared to bound-bound absorption in the visible regime, but is a significant contribution in the uv regime.

C. Free-Free Transitions

Free-free transitions occur when a photon is absorbed by an electron moving in the field of an ion or neutral, thereby increasing the energy of the electron. For the case of inverse Bremsstrahlung involving ions Kramer's formula modified by a Gaunt factor [8] provides a good estimate for the "cross section". Specifically, we use

$$\sigma_{ff}^+ = 2.43 \times 10^{-37} \frac{z^2}{T_e^{1/2}} \frac{\langle g_{ff} \rangle}{(h\nu)^3} \frac{\text{cm}^5}{\text{ion-elect.}} \quad (5)$$

where z is the net charge on the ion, T_e the temperature in e.v., $h\nu$ the photon energy in e.v., and $\langle g_{ff} \rangle$ the Gaunt factor averaged over a Maxwellian electron velocity distribution. In the field of N^+ this factor varies with temperature between 1.29 and 1.36; we have chosen an average value of 1.325. In the presence of N^{++} it varies between 1.175 and 1.26; here we use $\langle g_{ff} \rangle = 1.225$.

Free-free transitions in the fields of the neutrals are accounted for by the formula of Mjolsness and Ruppel [9]. This simple approximation, also used by Johnston et al. [3], is

$$\sigma_{ff}^0 = 2.20 \times 10^{-39} T_e^{3/2} \frac{\sigma_0}{(h\nu)^3} \left(1 + \frac{h\nu}{2T_e} \right) \frac{\text{cm}^5}{\text{neu.-elect.}} \quad (6)$$

where $\sigma_0 = .80$ for N. Geltman [10] has tabulated the absorption coefficients and cross sections for free-free radiation in electron-neutral collisions for different atomic species over a temperature range of 500 - 20000 °K and wavelengths between .5 and 20 μm . Eq. (6) gives results which differ from his by a factor of 2 at most.

For comparison with bound-bound and bound-free cross sections, eqs. (5) and (6) must be multiplied by Ne. On a log-log scale, both show essentially linearly decreasing behavior with increasing photon energy over the range of interest. The neutral cross section is two orders of magnitude less than that due to N^+ . At low temperatures the free-free cross sections are negligible compared to the others. However, at high temperatures and low photon energies they are of the same magnitude as the bound-free cross sections. As shown in section III these processes contribute significantly to the total absorption coefficient.

Finally, the total cross sections are shown in Figs. (1c) - (3c). Their behavior reflects the above discussion.

III. ATOMIC ABSORPTION COEFFICIENTS

The absorption coefficient, corrected for stimulated emission and defined as a function of photon frequency, is generally partitioned according to the various contributing processes, i.e.

$$\kappa_v = \kappa_{bb} + \kappa_{bf} + \kappa_{ff}^0 + \kappa_{ff}^+ \quad (\text{cm}^{-1}) \quad (7)$$

where the individual bound-bound, bound-free, and free-free coefficients are given by

$$\kappa_{bb} = \sum_{z, l} N_l^{z-1} \sum_u \sigma_{ul}^{z-1} \quad (8)$$

$$\kappa_{bf} = \sum_{z, l} N_l^{z-1} \sum_u \sigma_{ul}^{z-1} \quad (9)$$

$$\kappa_{ff}^0 = N \text{ Ne } \sigma_{ff}^0 \quad (10)$$

$$\kappa_{ff}^+ = N^+ \text{ Ne } \sigma_{ff}^+ \quad (11)$$

In equations (8) - (11) N_l^{z-1} , Ne, N, and N^+ are the population density for nitrogen in state l and ionization stage $z-1$, the total electron density, the total neutral density, and the total N^+ density, respectively. The cross sections are those discussed in the previous section. In eq. (8), u denotes the final state of the atom or ion in the same ionization stage after absorption, while in eq. (9) it denotes the state of the ion in the next higher ionization stage.

A. Plasma Composition

The detailed calculation of the absorption coefficient requires the population densities of all species in all states. In the present calculation we assume an initial neutral nitrogen density comparable to that found in equilibrium air [2,11] at the temperatures of interest. For an initial electron density of 10^{17} /cm³, the detailed rate equations given in reference [1] are integrated until a steady state is reached. The final populations of all states contained within our model are then used to calculate the absorption coefficients given above. Tables I and II show a comparison between the final N, N⁺, N⁺⁺, and Ne densities for our model plasma (I) and the corresponding contribution to equilibrium air (II); a state-specific decomposition is not shown.

B. Absorption Coefficients

The total absorption coefficients, defined in eq.(7), are shown in Figures (1d) - (3d). Explicit representations of the bound-bound, bound-free, and free-free contributions are also shown specifically, in Figures (1e-1h) - (3e-3h).

At $T_e = 1.0$ e.v. the absorption coefficient shows detailed structure in the ir and near uv regime due to significant contributions from neutral nitrogen bound-bound transitions. The absorption length for the uv bound-bound radiation is approximately 10^{-6} to 10^{-3} cm while the ir radiation has a length of 10^{-2} to 10^{-1} cm. Bound-free transitions contribute predominately in the visible regime; the absorption length here is on the order of 100 cm. At this temperature the contribution of free-free transitions involving neutrals is nearly equal to that involving ions. This is because the number of neutrals exceeds the number of ions by exactly two orders of magnitude. Free-free absorption only plays a noticeable role for low photon energies (ir - visible).

When the electron temperature and density of the plasma is increased, the total number of ions becomes comparable to the number of neutrals (see Table I). Figures (2d) - (2h) show several effects. In the ir regime the absorption coefficient reflects the changes in the bound-bound cross sections, particularly broader line structure. While free-free transitions in the presence of neutrals contributes very little, the increased number of ions and electrons causes free-free transitions in the ion fields to significantly enhance the absorption coefficient in the ir and visible regime. More highly excited N states also

contribute to an increase in κ in the visible regime, through bound-free transitions. In the uv regime there is a noticeable increase in N^+ bound-bound contributions. As before, most uv radiation is absorbed through bound-free transitions.

Finally, the results for $T_e = 3.0$ e.v. are shown in Figures (3d) - (3h). Here the plasma consists largely of N^+ and electrons. In the ir and visible range, the absorption coefficient has increased by approximately a factor of two, largely due to free-free absorption. Bound-bound absorption involving neutrals has less of an effect than that involving ions. A decrease in nitrogen atoms in the ground state configurations has resulted in a corresponding decrease in bound-free uv absorption involving these atoms.

As mentioned earlier, Johnston et. al. [3] have calculated the total absorption coefficient in equilibrium air for a series of temperatures. Their calculations include thousands of bound-bound and bound-free transitions for the dominant atomic and molecular species at each temperature. We have compared the results of our simple model calculation to their results for temperatures of 1.5 and 2.0 e.v.. At these temperatures, air consists largely of nitrogen and oxygen atoms and their ions. The nitrogen concentrations are similar to those in our model (see

Tables I and II). For photon energies between 1.7 and 100.0 e.v., the general agreement is very good. Excluding the detailed line structure, our absorption coefficient is within a factor of 1.1 - 2.0 throughout. Even the detailed line structure is in reasonable agreement for energies between 1.6 and 20.0 e.v.. Certainly their computations should show more bound-bound contributions due to the inclusion of a complete description of oxygen, as well as more nitrogen transitions. The largest discrepancy is in the ir region, energies less than 1.6 e.v., where our calculations show structure in contrast to theirs, although the average coefficient there still agrees within a factor of three.

Table I

Model Plasma Composition ($/\text{cm}^3$)*

<u>Te(e.v)</u>	<u>N^{Total}</u>	<u>N</u>	<u>N⁺</u>	<u>N⁺⁺</u>	<u>Ne</u>
1.0	3.56×10^{19}	3.52×10^{19} .99	3.62×10^{17} .01	9.39×10^8	3.62×10^{17}
1.5	3.90×10^{19}	3.23×10^{19} .83	6.72×10^{18} .17	2.88×10^{13}	6.72×10^{18}
2.0	3.93×10^{19}	1.89×10^{19} .48	2.03×10^{19} .52	6.65×10^{15}	2.03×10^{19}
2.5	3.92×10^{19}	8.69×10^{18} .22	3.04×10^{19} .77	1.65×10^{17}	3.07×10^{19}
3.0	3.93×10^{19}	3.90×10^{18} .10	3.40×10^{19} .87	1.41×10^{18} .04	3.68×10^{19}

* Bottom entries give fraction of total nitrogen composition

Table II

Equilibrium Air Nitrogen Composition ($/\text{cm}^3$)*

<u>Te(e.v.)</u>	<u>N^{Total}</u>	<u>N</u>	<u>N⁺</u>	<u>N⁺⁺</u>	<u>Ne</u>
1.0	3.45×10^{19}	3.40×10^{19} .99	4.50×10^{17} .01	-	5.60×10^{17}
1.5	3.80×10^{19}	3.25×10^{19} .86	5.50×10^{18} .15	-	6.60×10^{18}
2.0	3.84×10^{19}	2.15×10^{19} .56	1.69×10^{19} .44	-	2.00×10^{19}
2.5	3.85×10^{19}	1.05×10^{19} .27	2.08×10^{19} .54	7.20×10^{18} .19	3.10×10^{19}
3.0	3.85×10^{19}	5.00×10^{18} .13	2.15×10^{19} .56	1.20×10^{19} .32	4.30×10^{19}

* Bottom entries give fraction of total nitrogen composition

IV. CONCLUDING REMARKS

The photoabsorption properties of a model nitrogen plasma have been studied for electron temperatures between 1.0 and 3.0 e.v.. Specifically, the relevant absorption cross sections for processes involving bound-bound, bound-free, and free-free transitions have been computed. Finally, the total and partial absorption coefficients (frequency-dependent) have been obtained for a nitrogen plasma whose composition is similiar to that of high temperature, full density, equilibrium air. The results of this benchmark calculation are consistent and in good agreement with the earlier numerical studies of Johnston et. al.. These absorption cross sections and coefficients provide the necessary input for studying radiative transfer in a nitrogen plasma, the next step in our ongoing program.

TE=1.0 NE=3.621E17

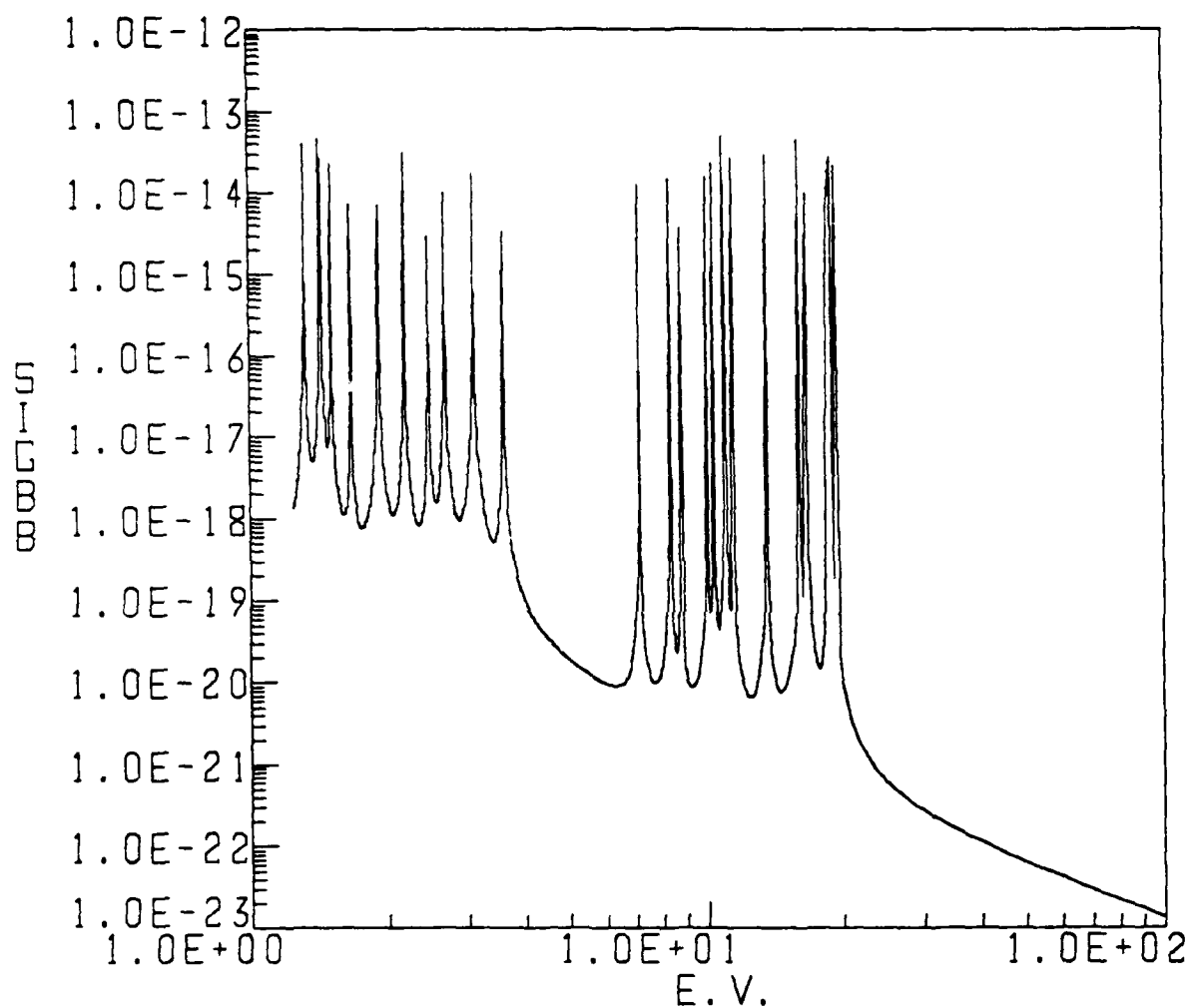


Fig. 1a — Photoabsorption cross section for bound-bound transitions. $T_e = 1.0\text{ e.v.}$

TE=1.0 NE=3.621E17

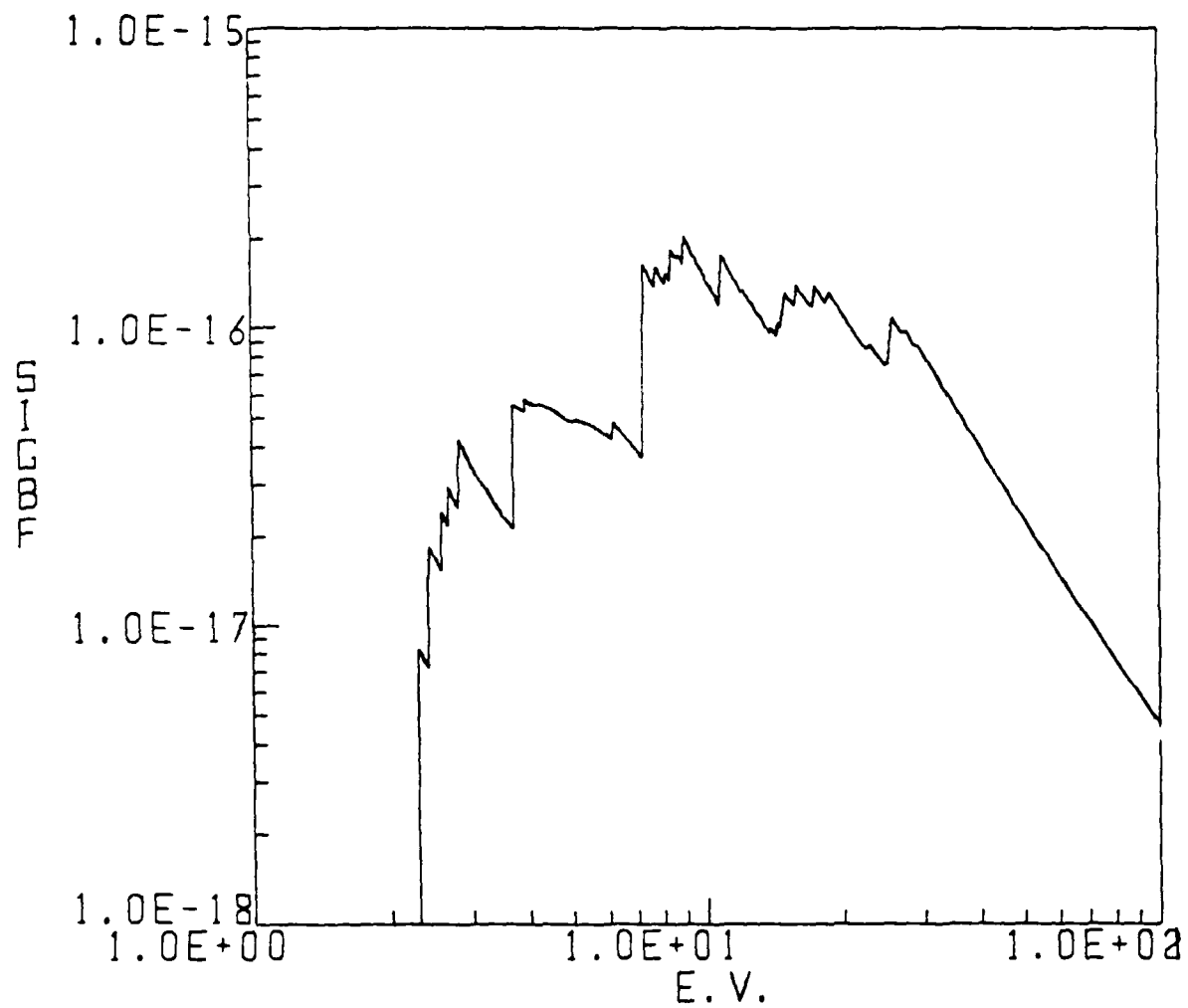


Fig. 1b — Bound-free cross section. $T_e = 1.0$ e.v.

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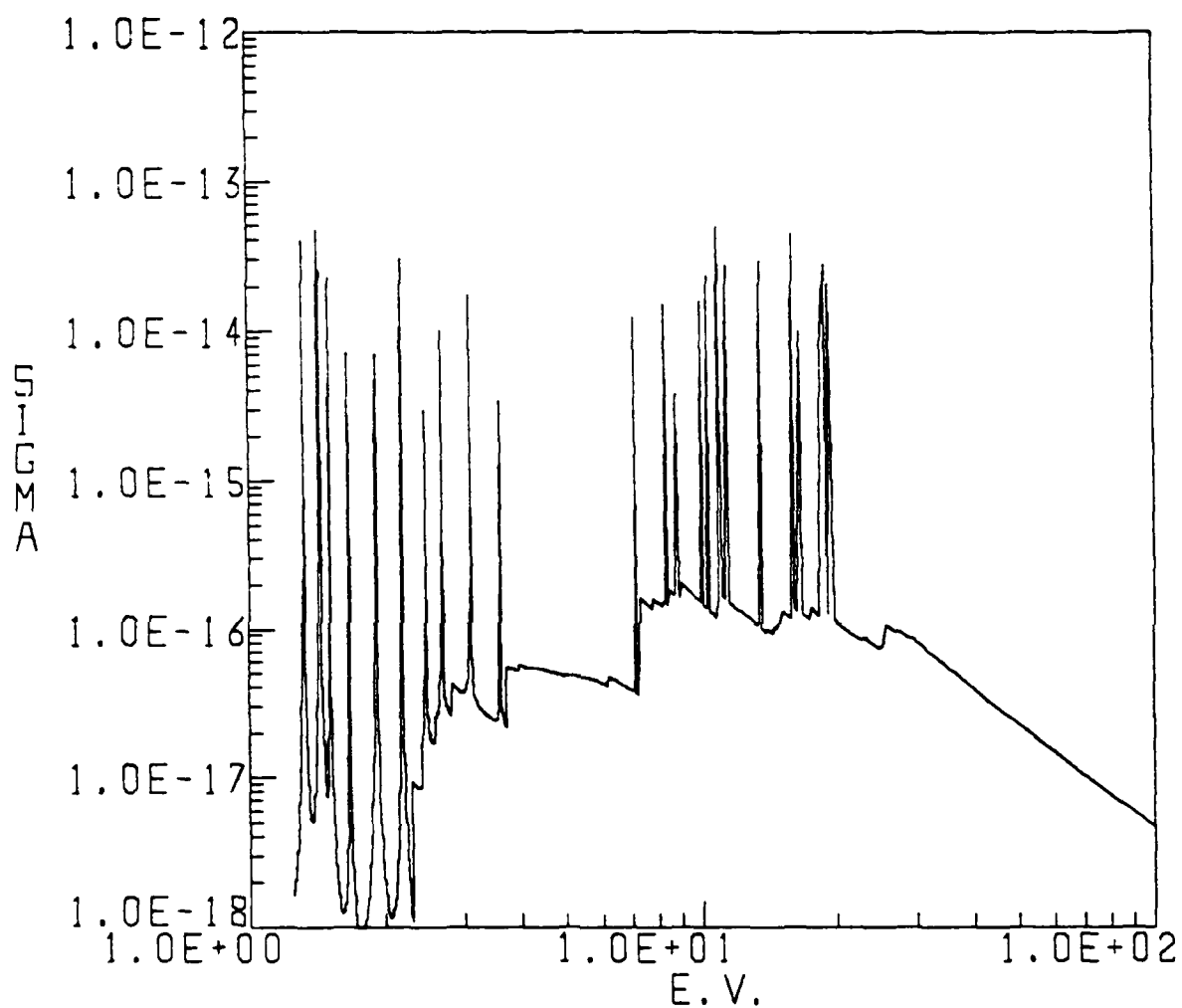


Fig. 1c — Total photoabsorption cross section. Te = 1.0 e.v.

TE=1.0 NE=3.621E17

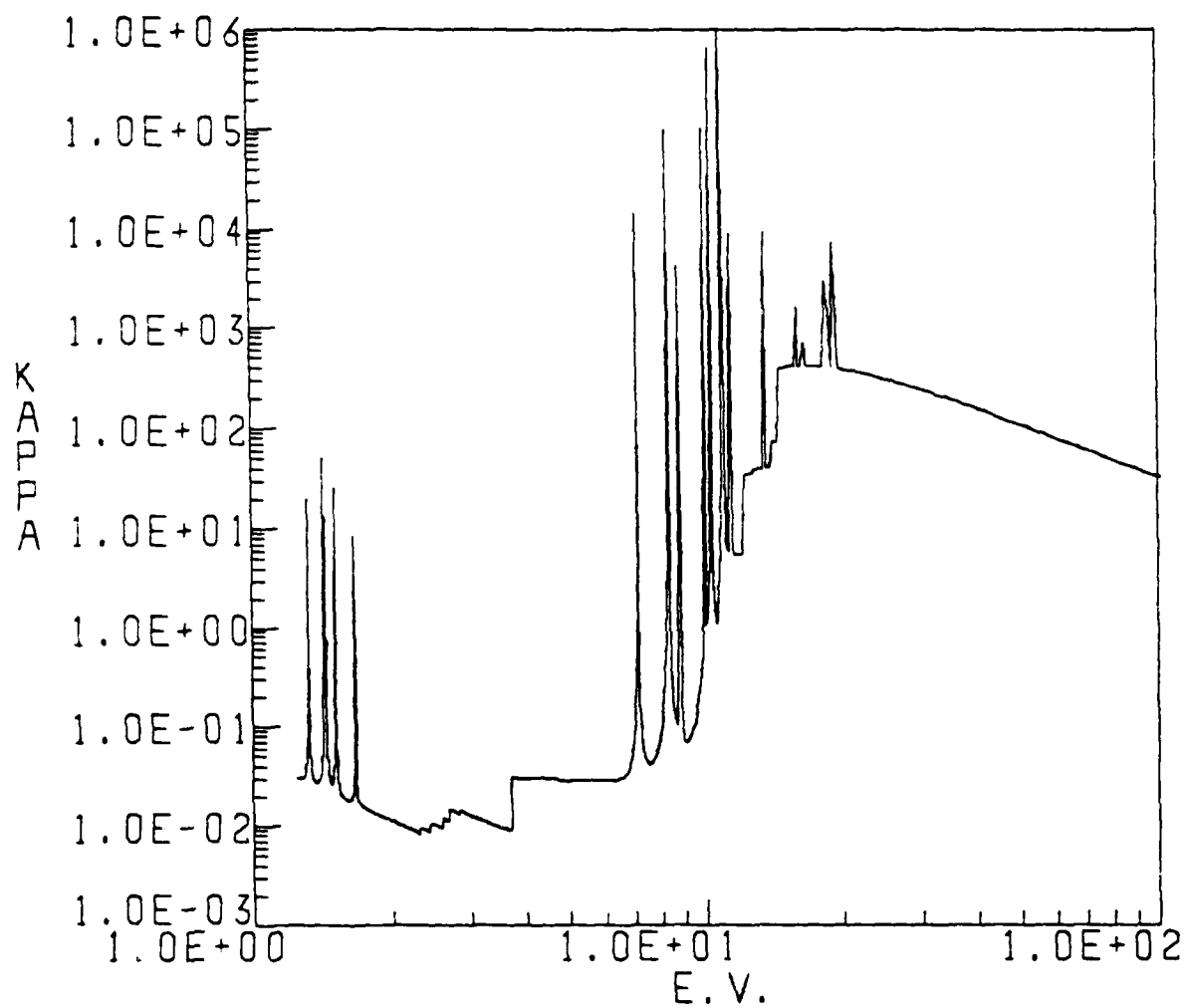


Fig. 1d — Total photoabsorption coefficient. $T_e = 1.0$ e.v.

TE=1.0 NE=3.621E17

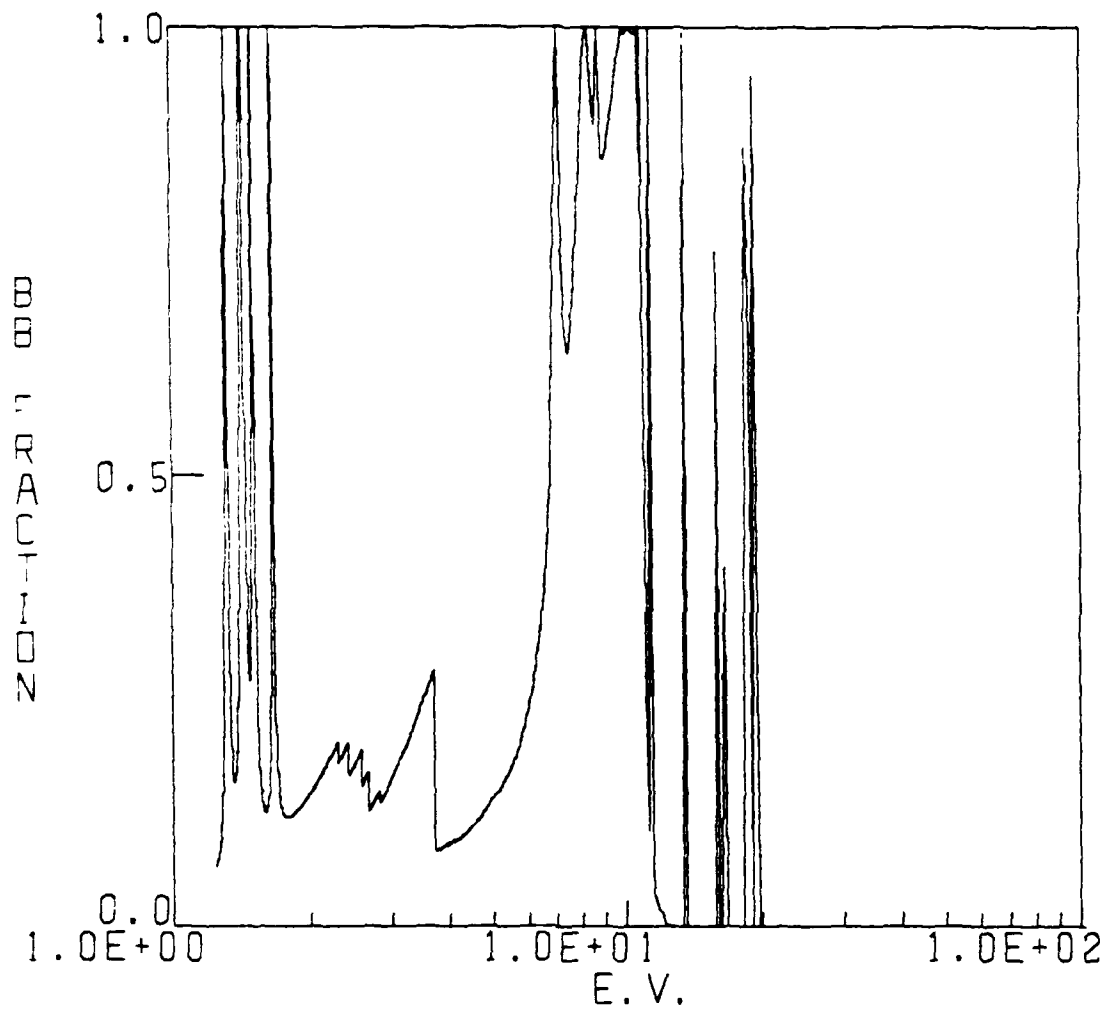


Fig. 1e — Fraction of the total photoabsorption coefficient due to bound-bound transitions. $T_e = 1.0$ e.v.

TE=1.0 NE=3.621E17

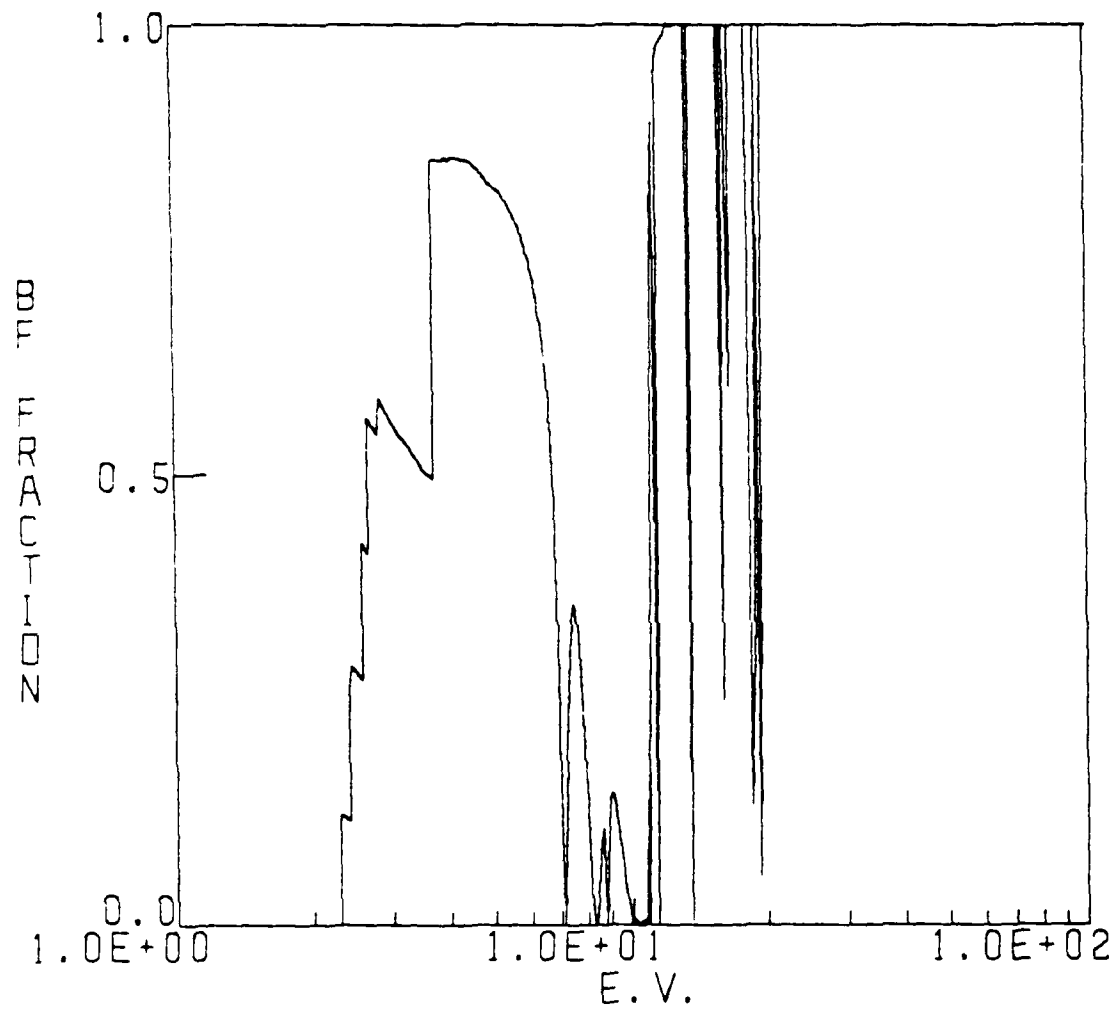


Fig. 1f — Fraction of the total photoabsorption coefficient due to bound-free transitions. $T_e = 1.0$ e.v.

TE=1.0 NE=3.621E17

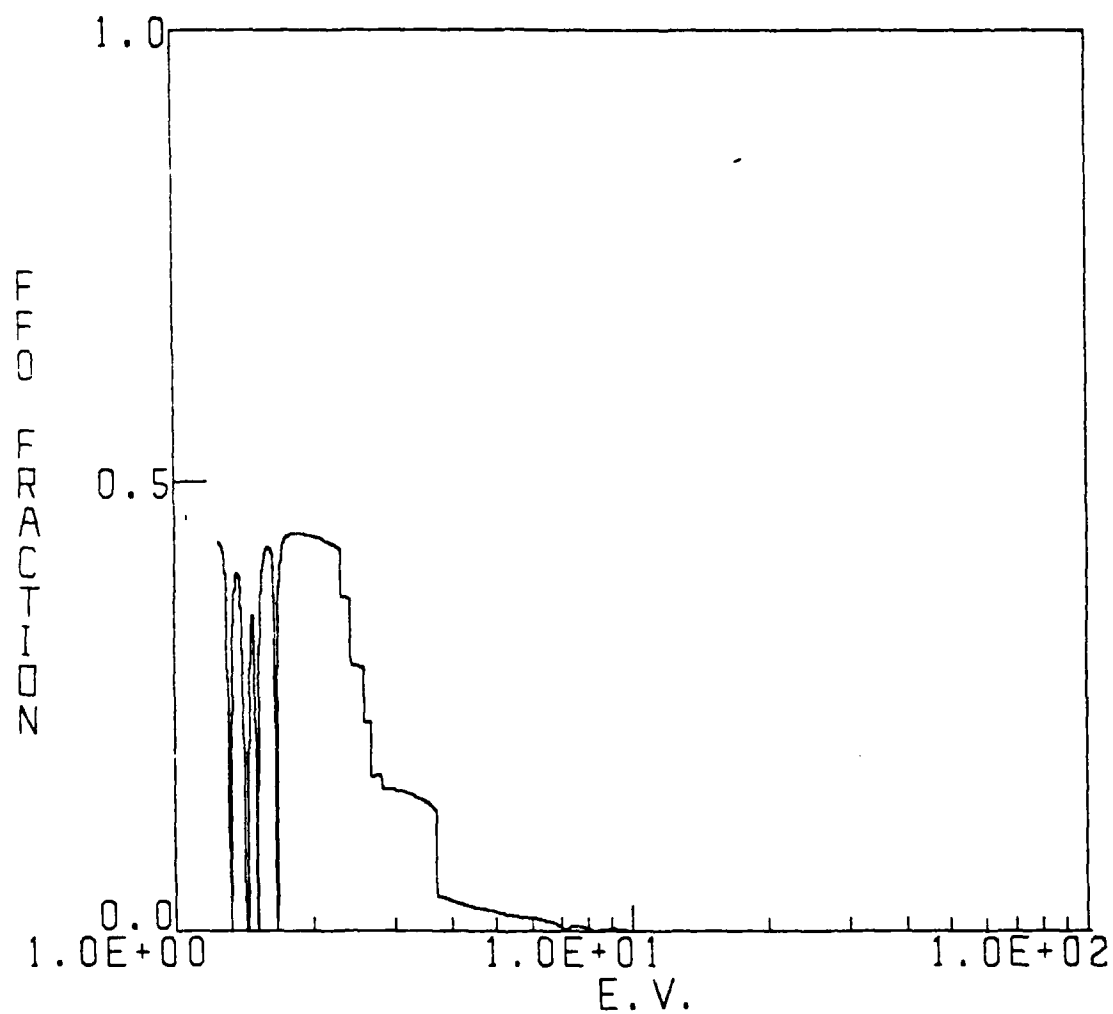


Fig. 1g — Fraction of the total photoabsorption coefficient due to free-free transitions involving neutrals. $T_e = 1.0$ e.v.

TE=1.0 NE=3.621E17

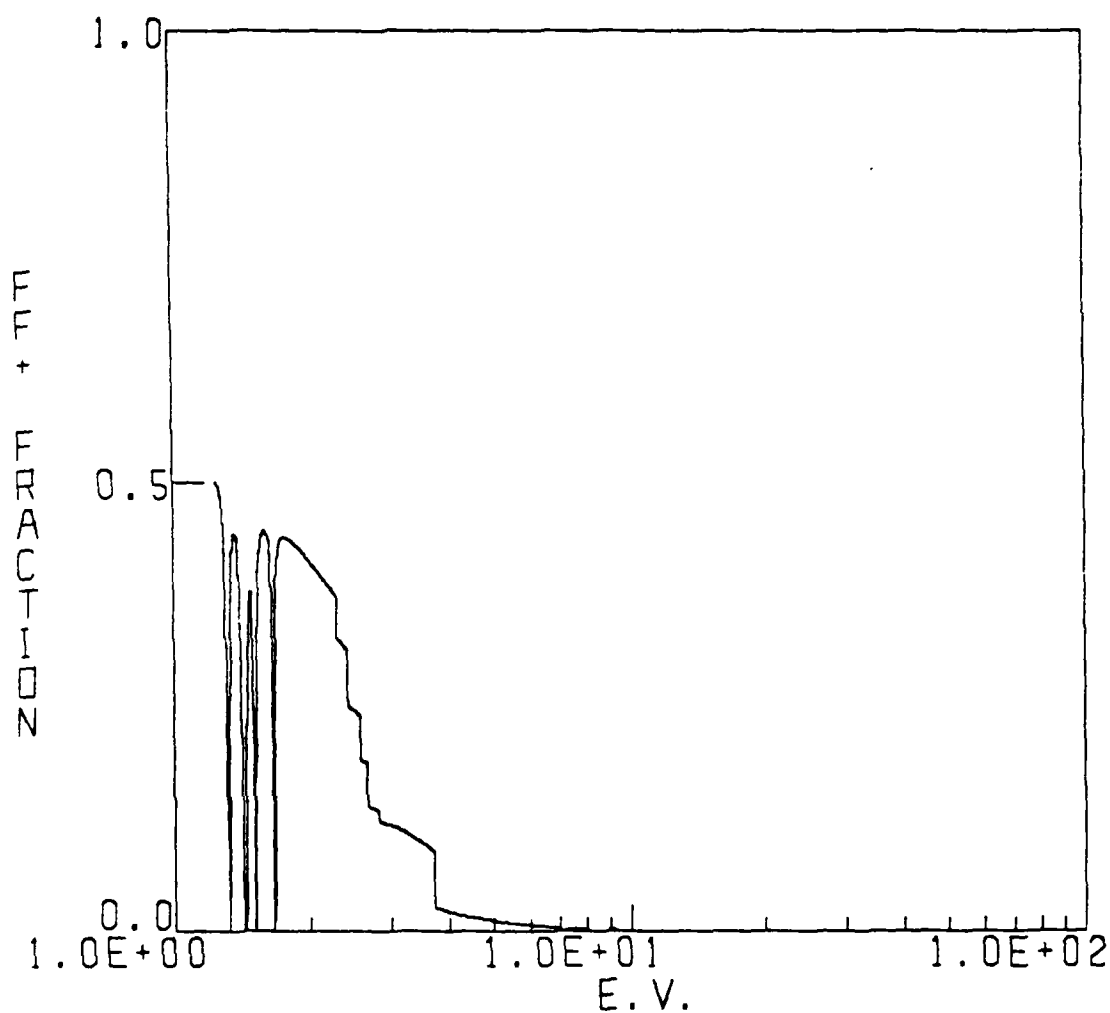


Fig. 1h — Fraction of the total photoabsorption coefficient due to free-free transitions involving ions. $T_e = 1.0$ e.v.

TE=2.0 NE=2.033E19

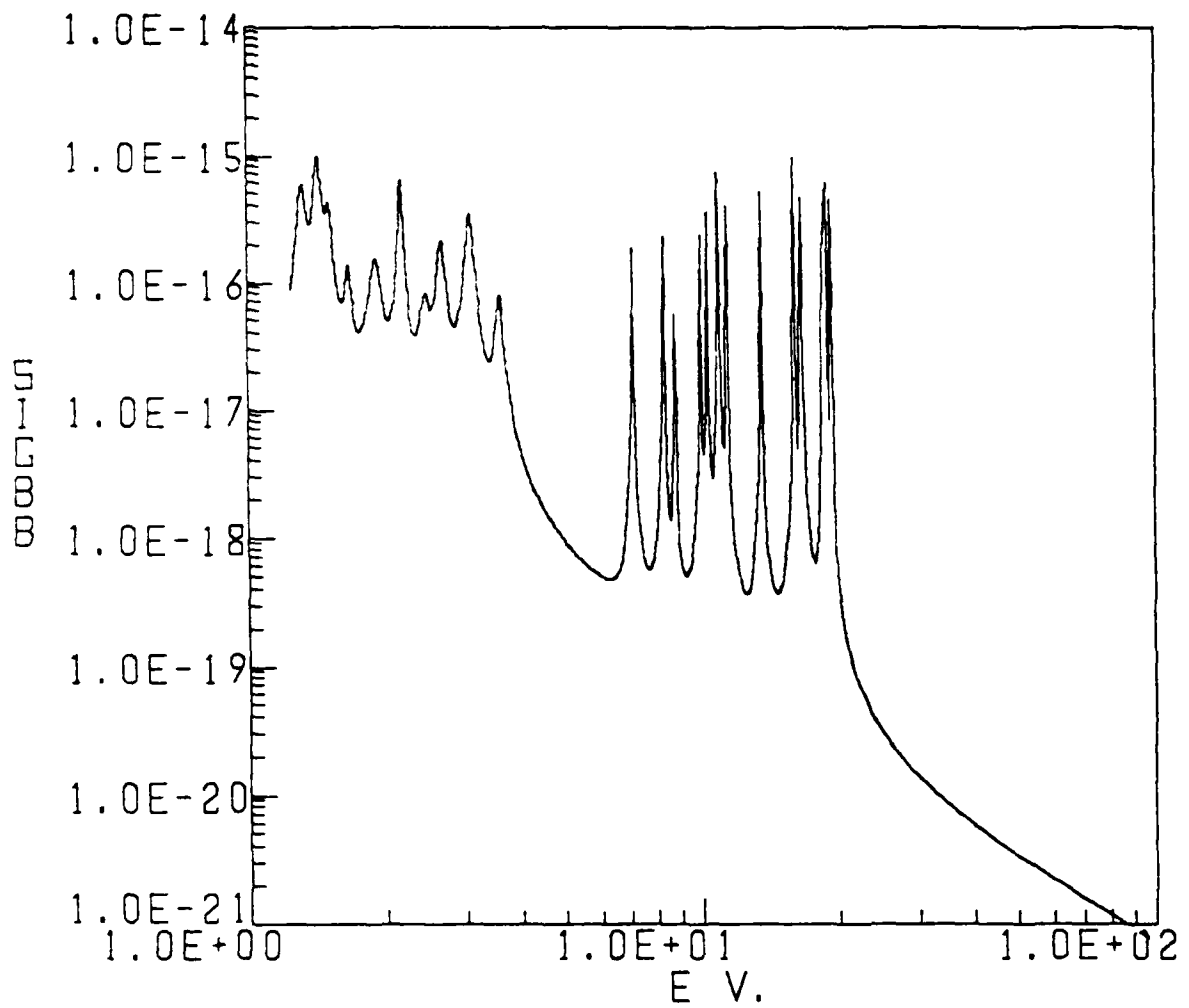


Fig. 2a — Same as Figure 1a. $T_e = 2.0$ e.v.

TE=2.0 NE=2.033E19

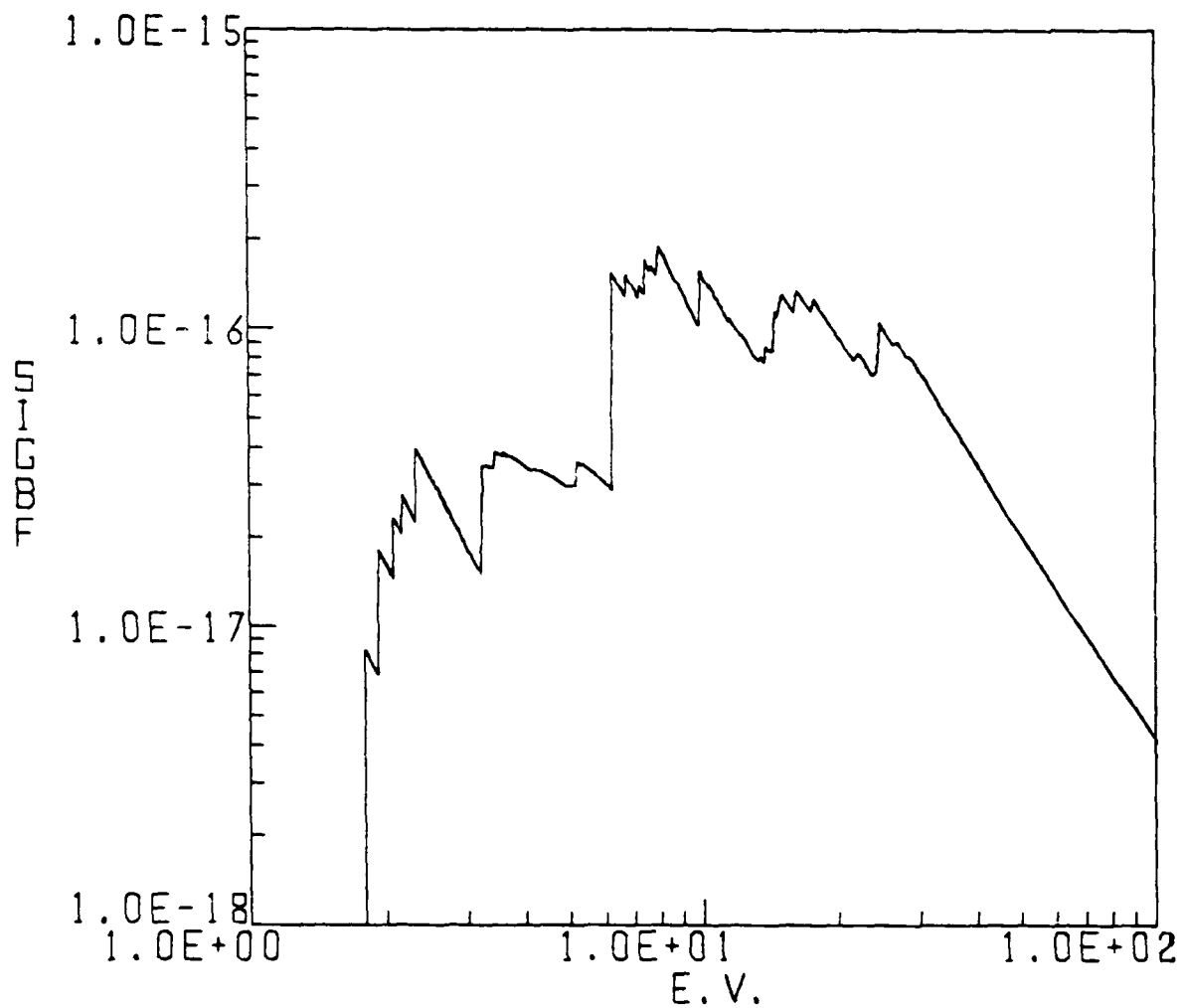


Fig. 2b — Same as Figure 1b. $T_e = 2.0$ e.v.

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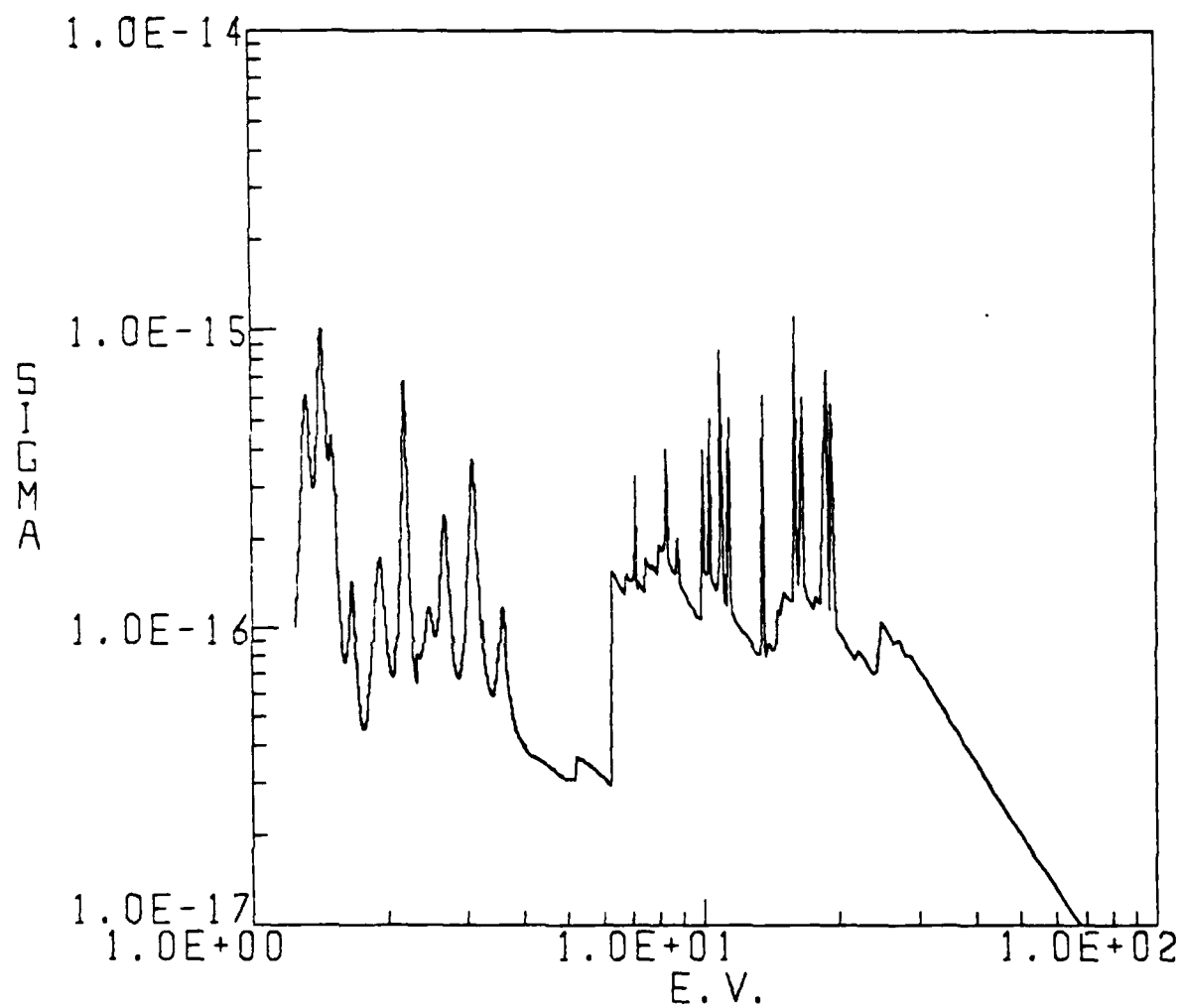


Fig. 2c — Same as Figure 1c. $T_e \approx 2.0$ e.v.

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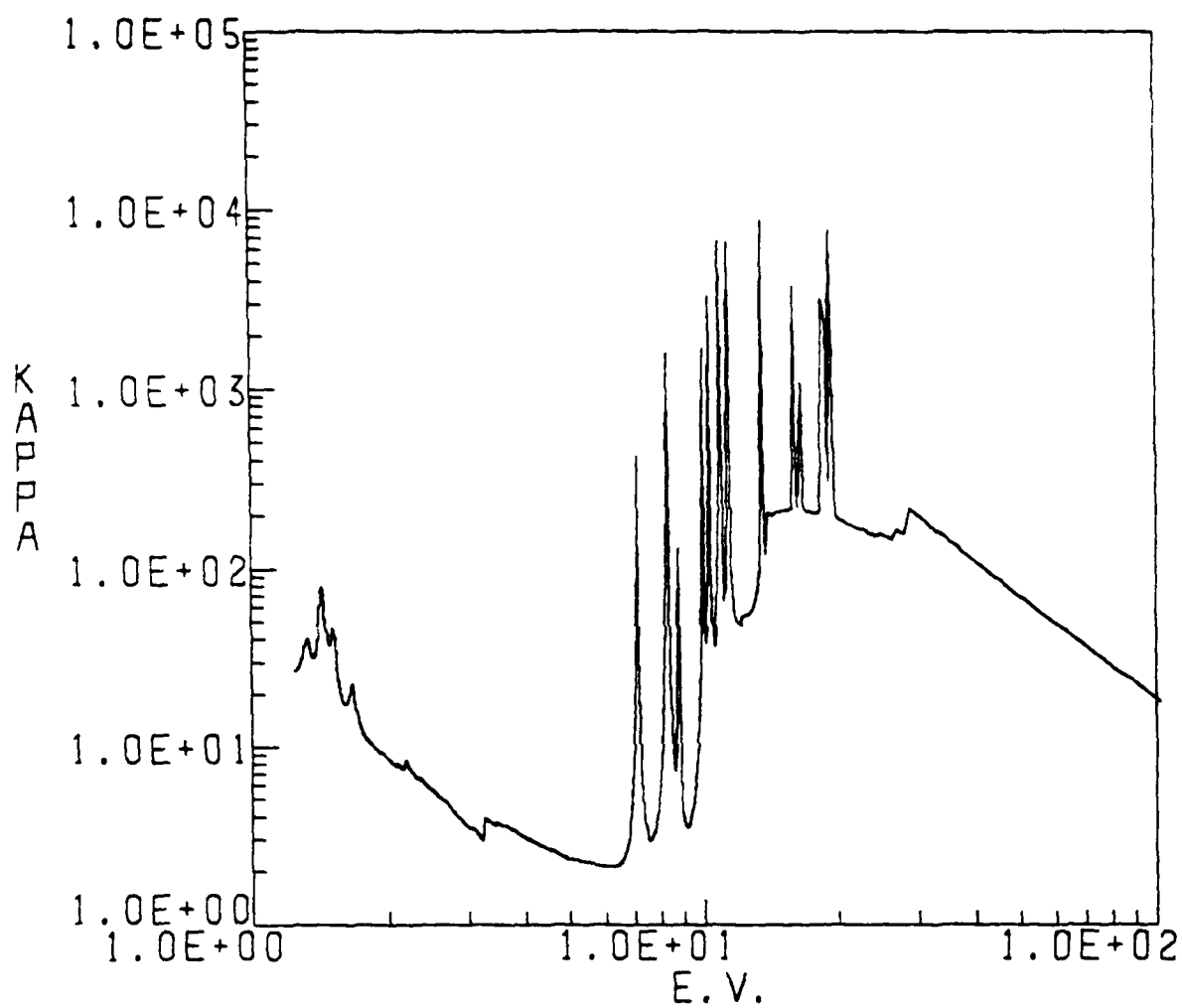


Fig. 2d — Same as Figure 1d. Te = 2.0 e.v.

TE=2.0 NE=2.033E19

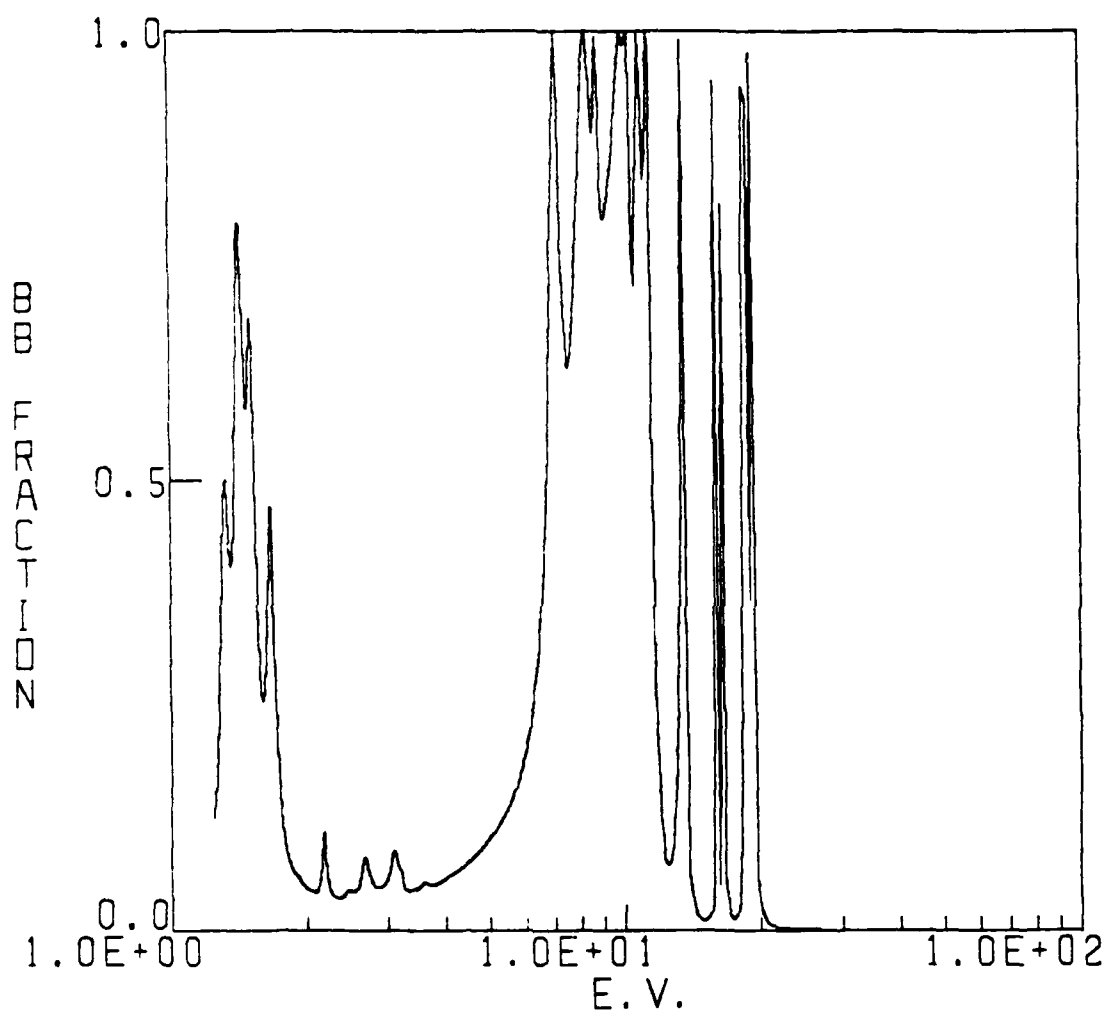


Fig. 2e — Same as Figure 1e. $T_e = 2.0$ e.v.

TE=2.0 NE=2.033E19

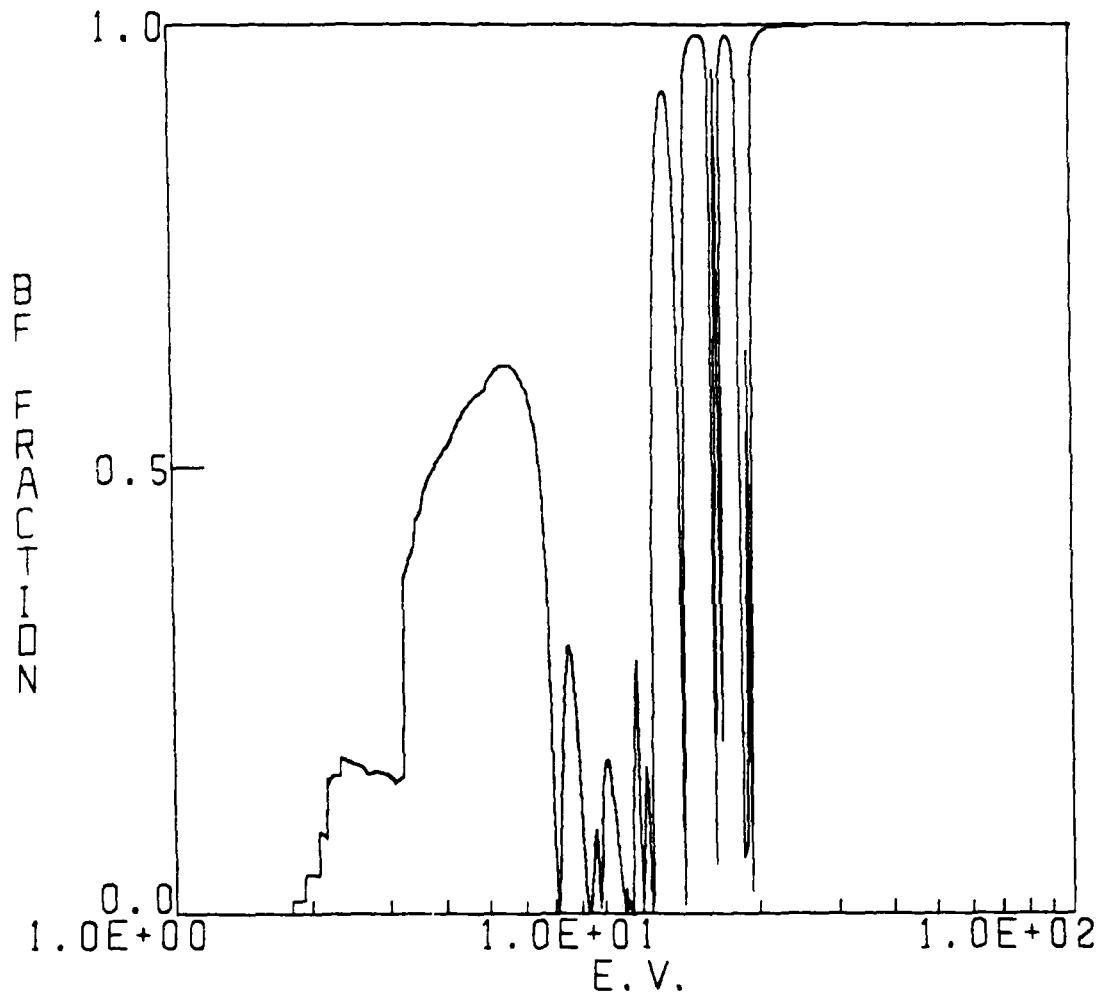


Fig. 2f — Same as Figure 1f. $T_e = 2.0$ e.v.

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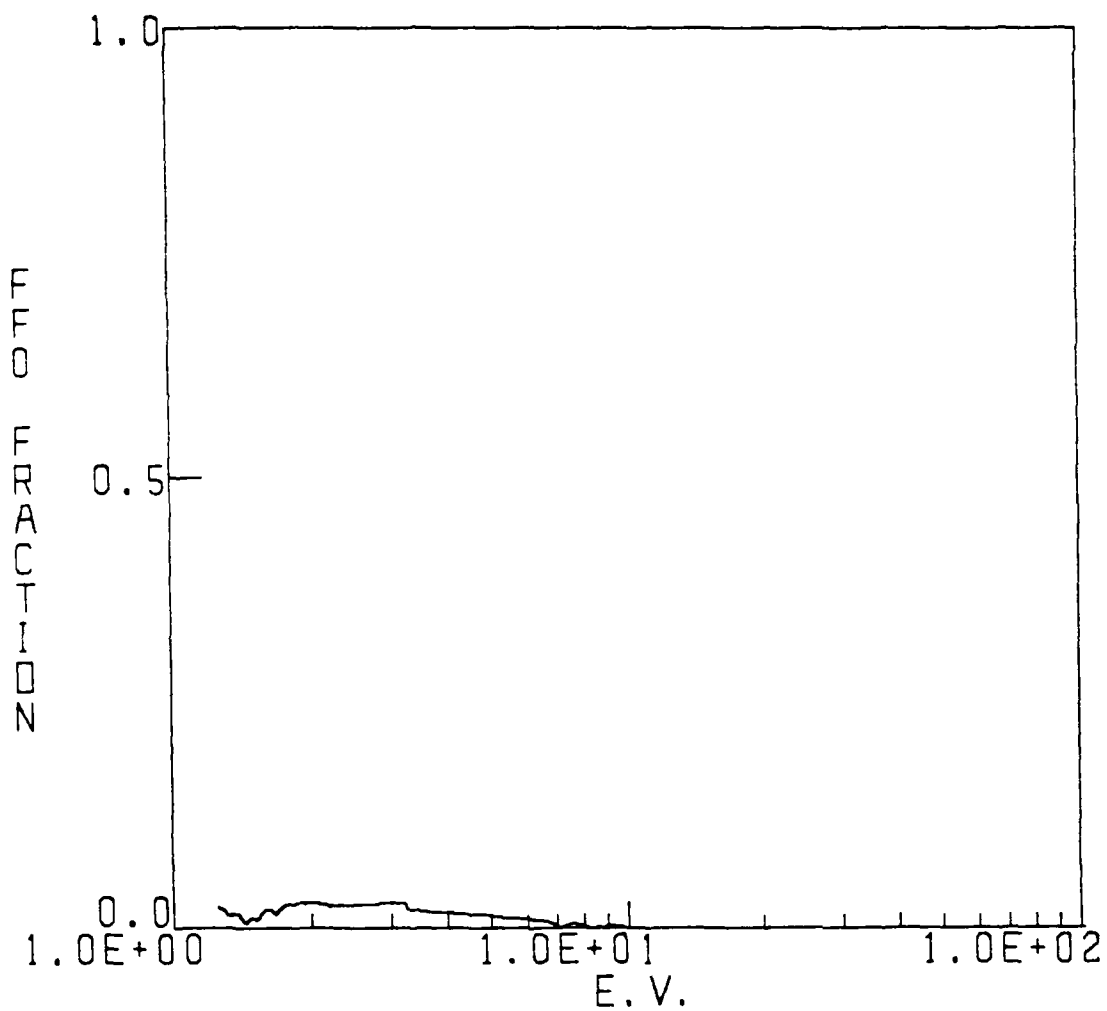


Fig. 2g — Same as Figure 1g. $T_e = 2.0$ e.v.

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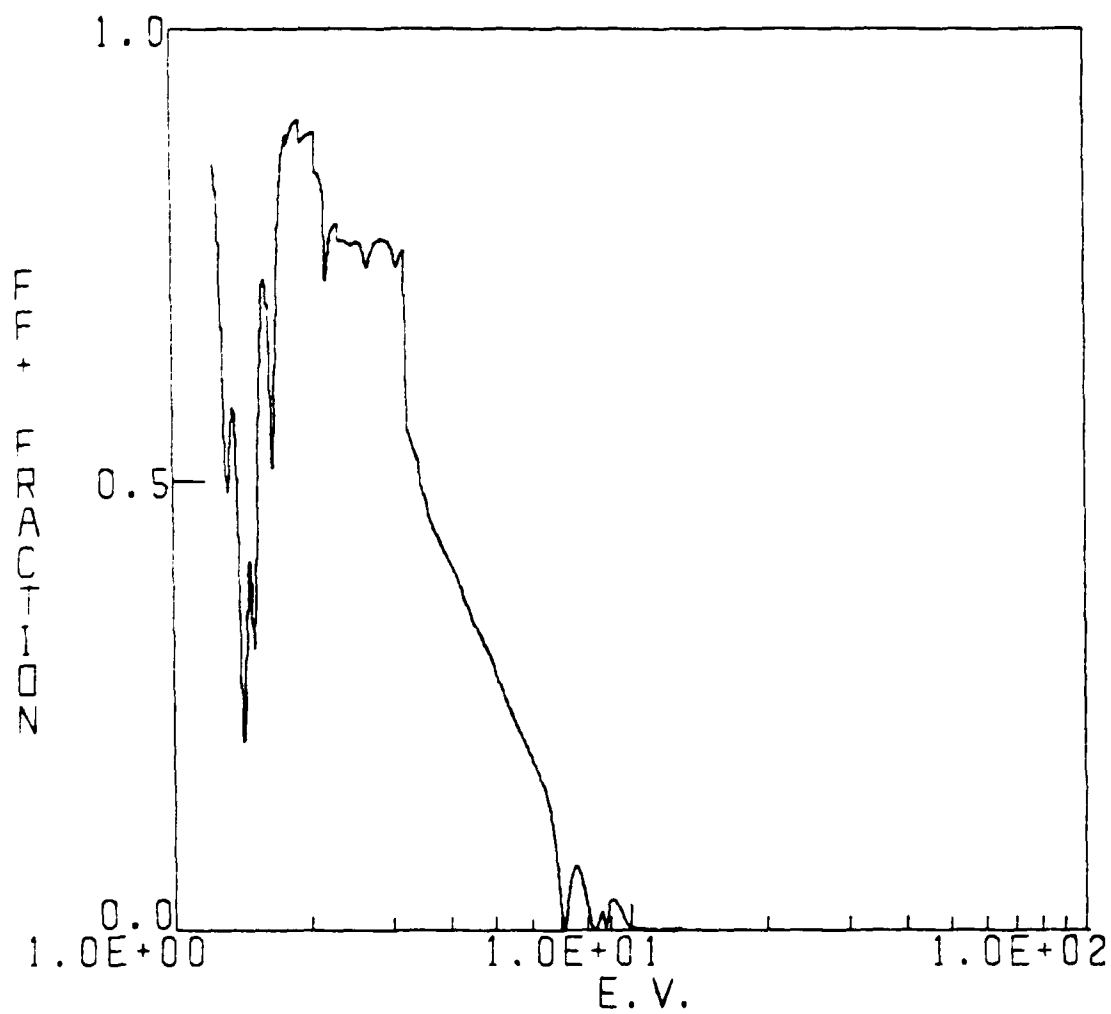


Fig. 2h — Same as Figure 1h. $T_e = 2.0$ e.v.

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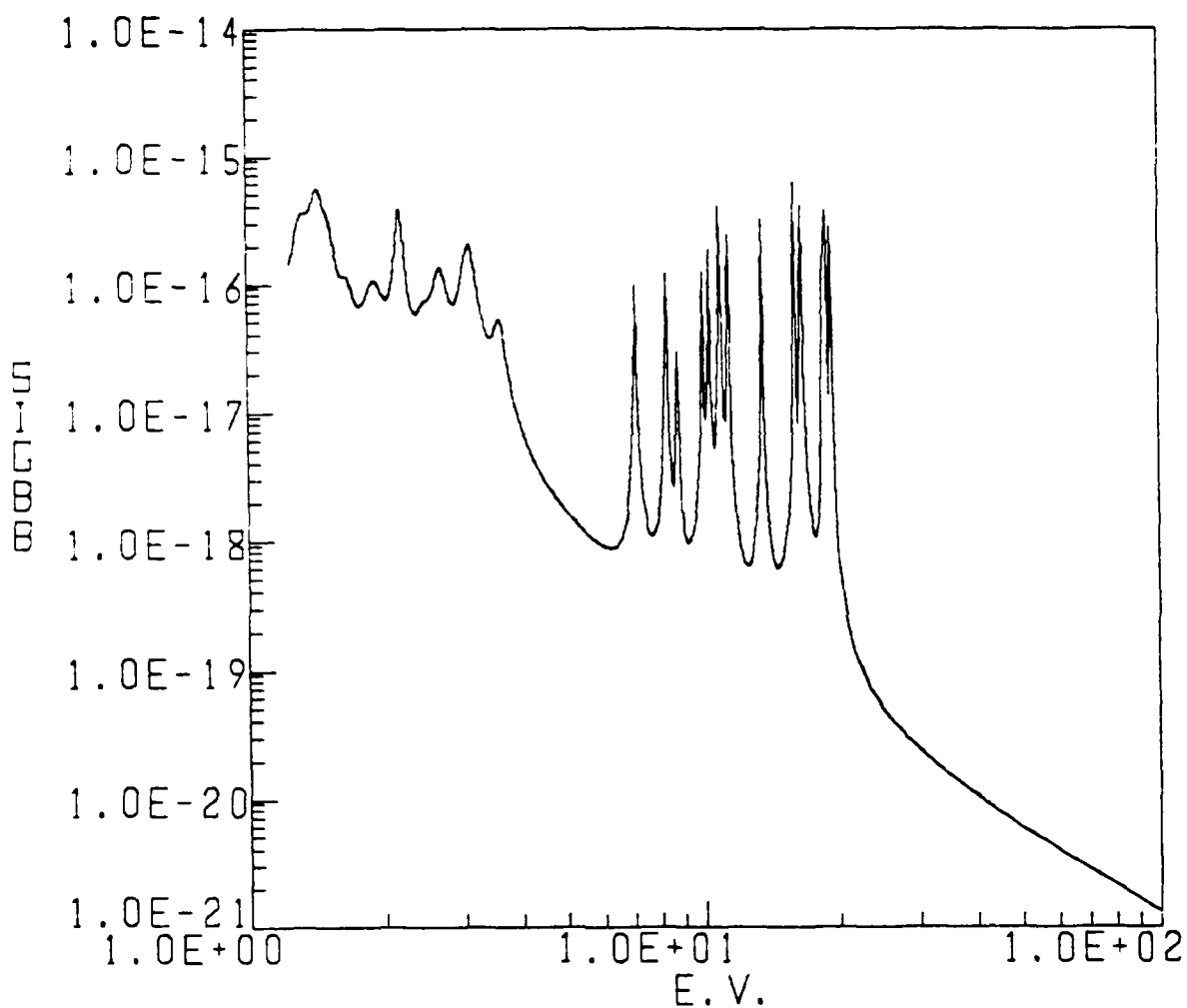


Fig. 3a — Same as Figure 1a. $T_e = 3.0$ e.v.

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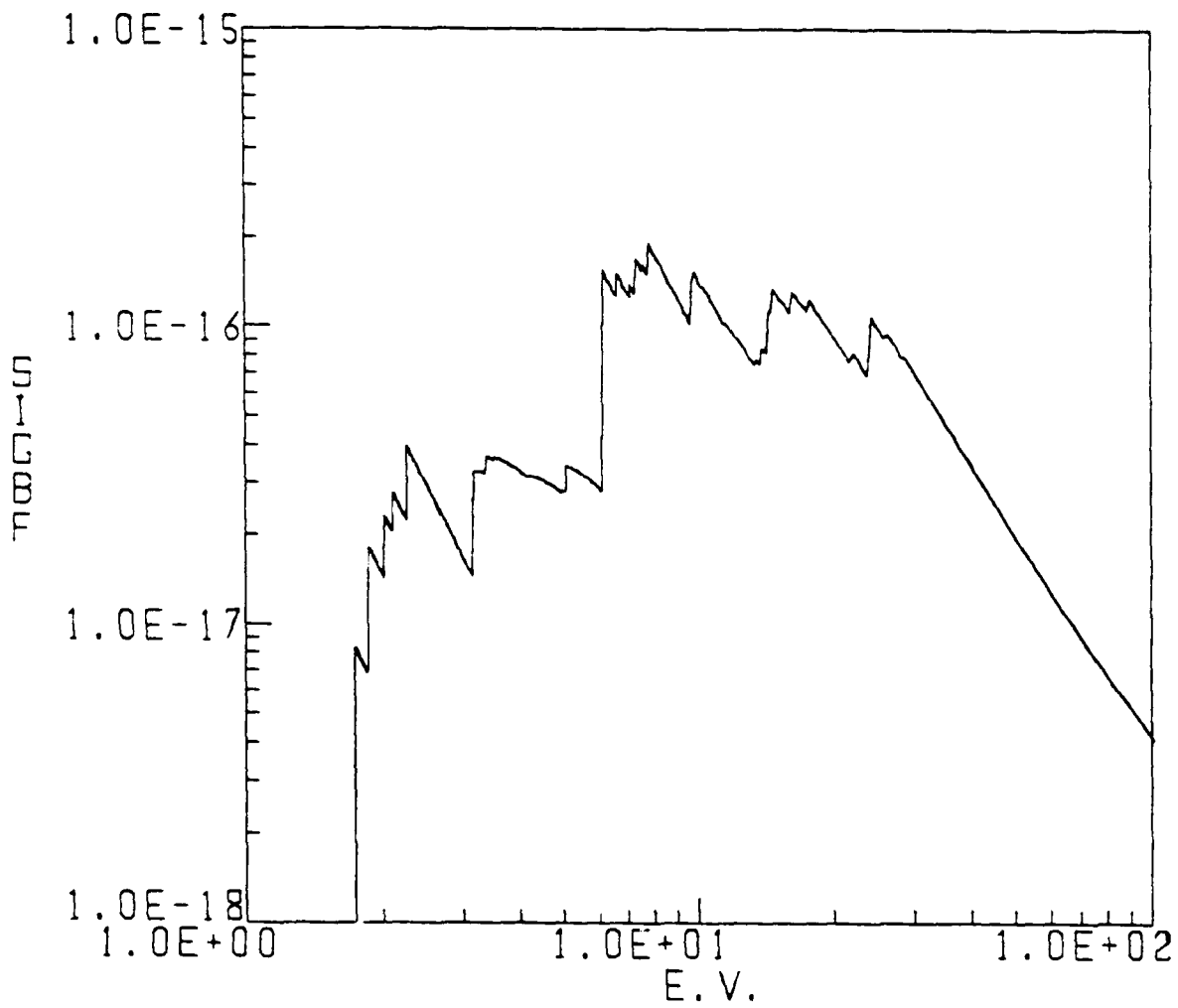


Fig. 3b — Same as Figure 1b. Te = 3.0 e.v.

TE=3.0 NE=3.677E19

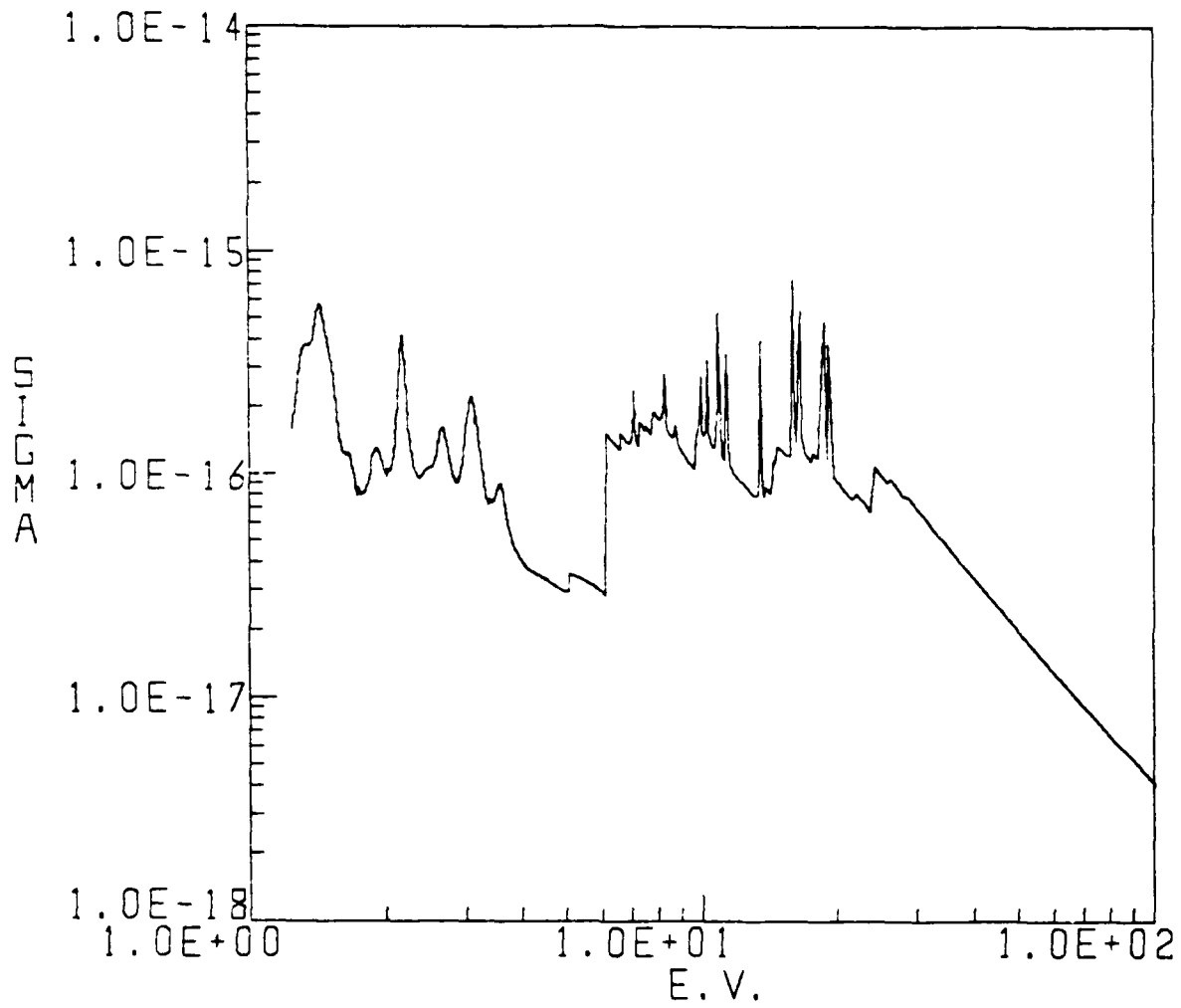


Fig. 3c — Same as Figure 1c. $T_e = 3.0$ e.v.

TE=3.0 NE=3.677E19

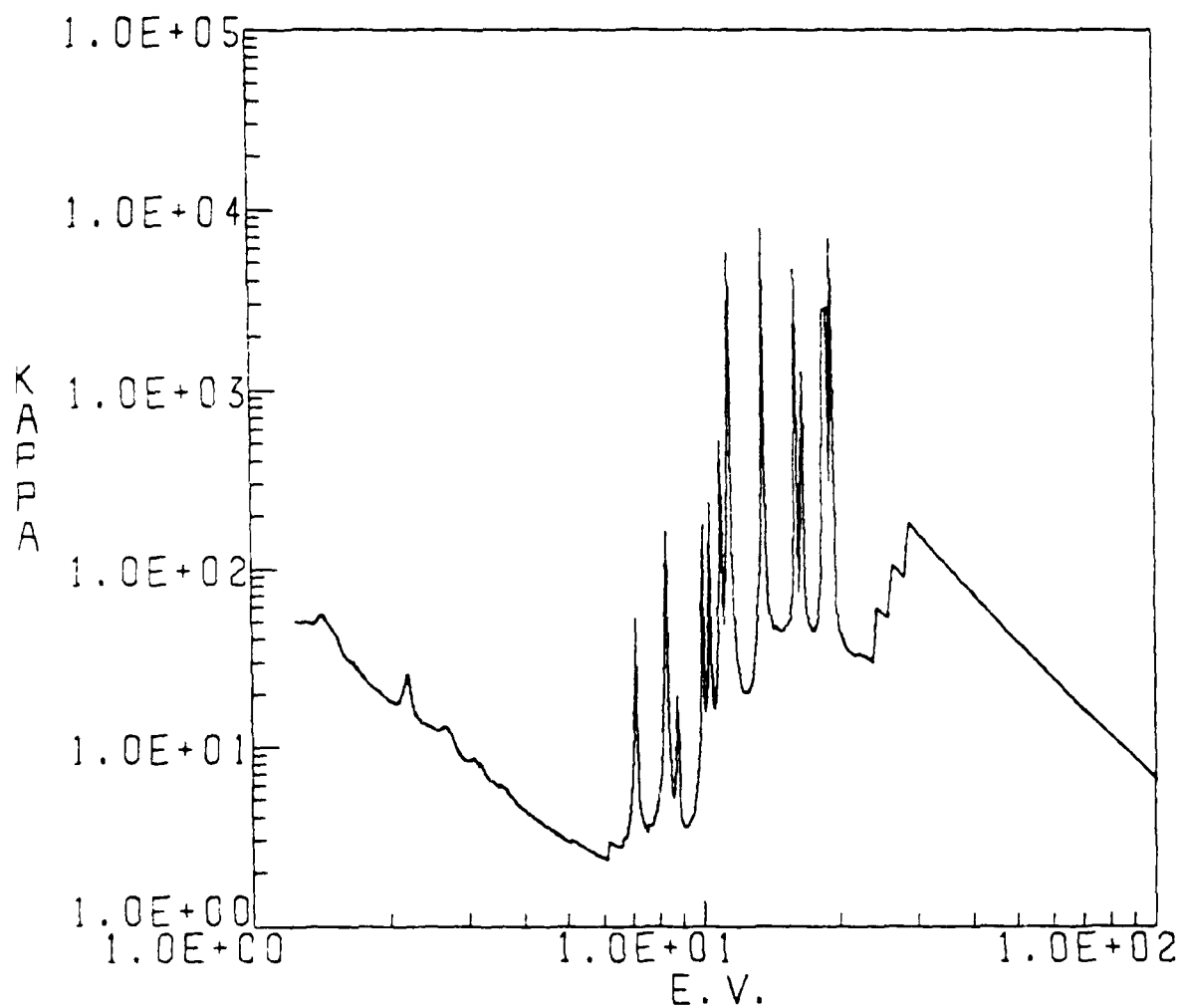


Fig. 3d — Same as Figure 1d. Te = 3.0 e.v.

TE=3.0 NE=3.677E19

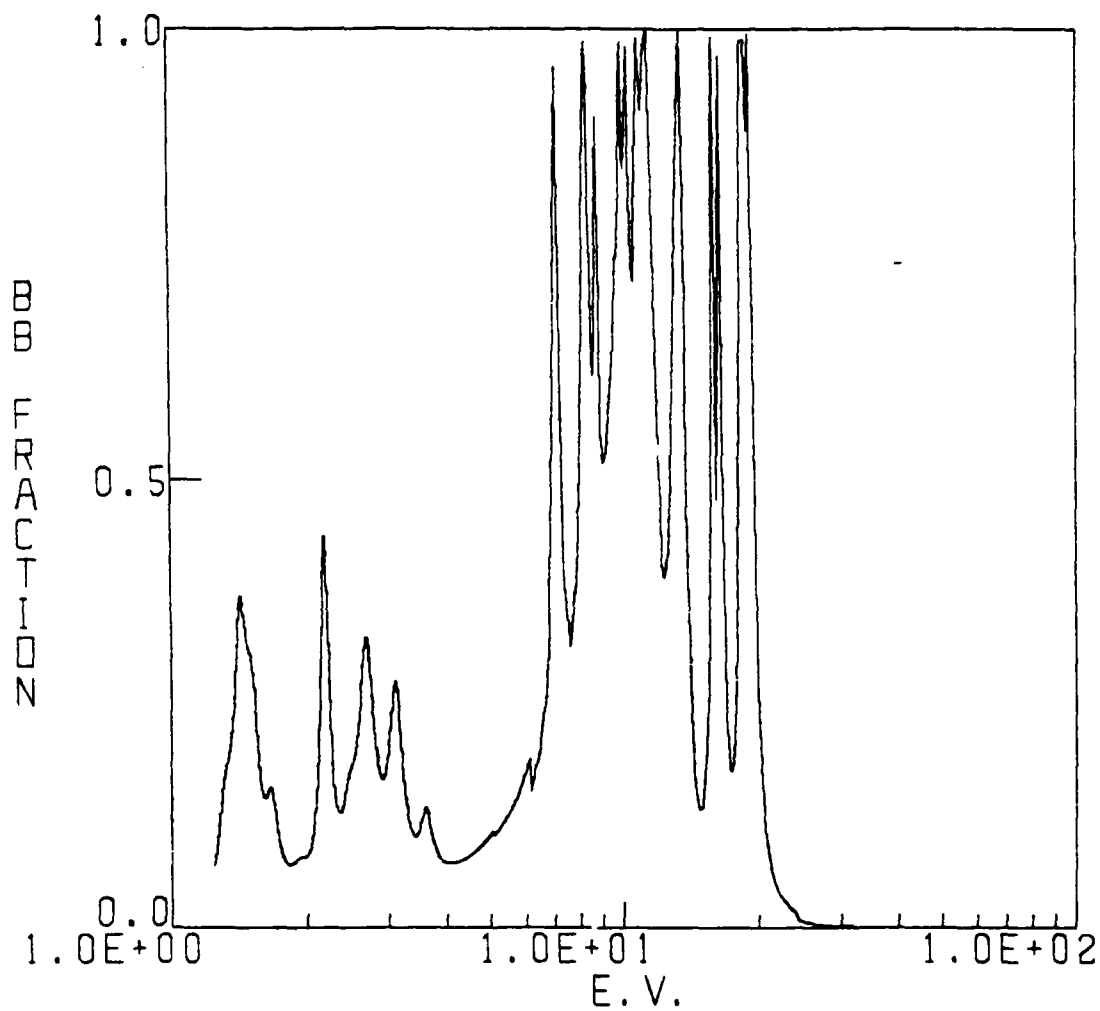


Fig. 3e — Same as Figure 1e. $T_e = 3.0$ e.v.

TE=3.0 NE=3.677E19

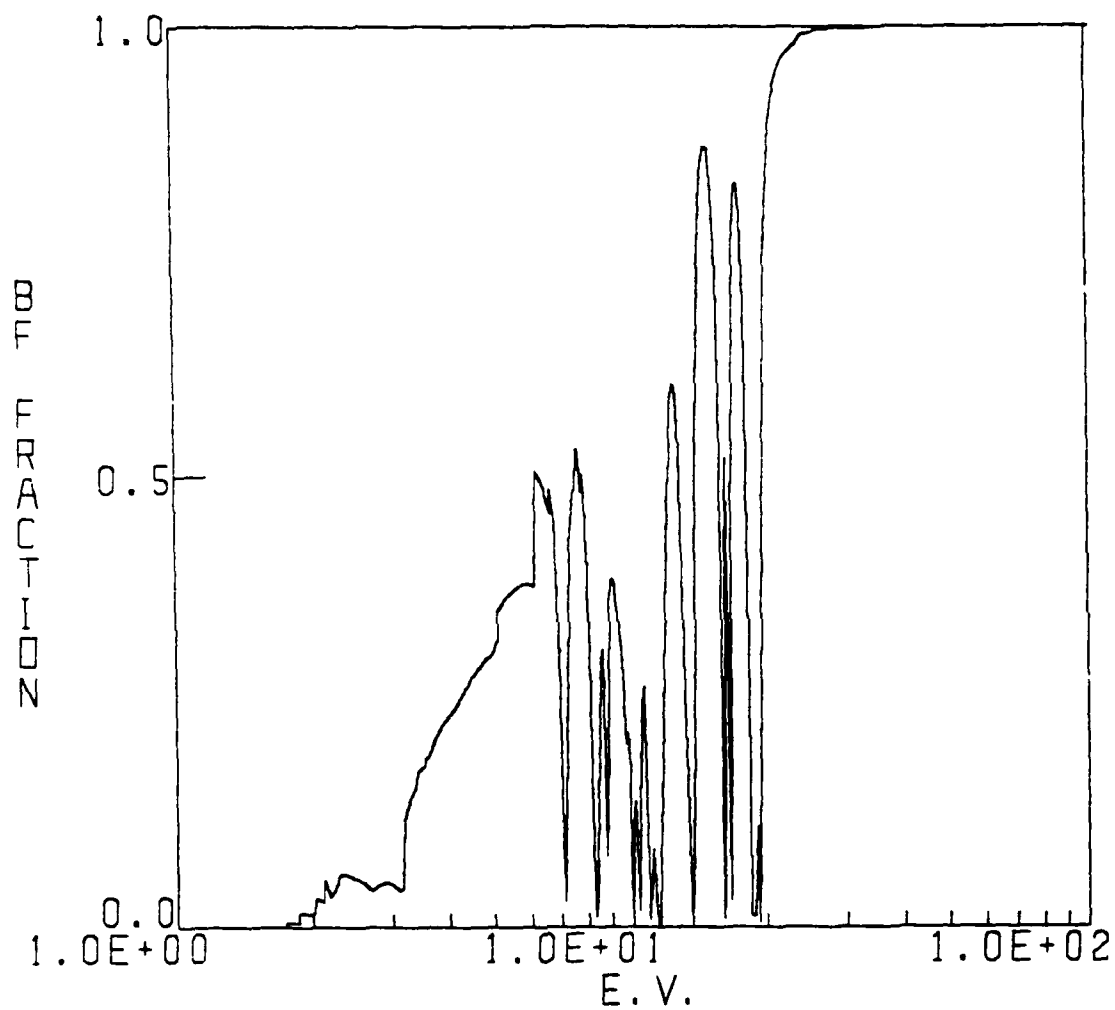


Fig. 3f — Same as Figure 1f. $T_e = 3.0$ e.v.

TE=3.0 NE=3.677E19

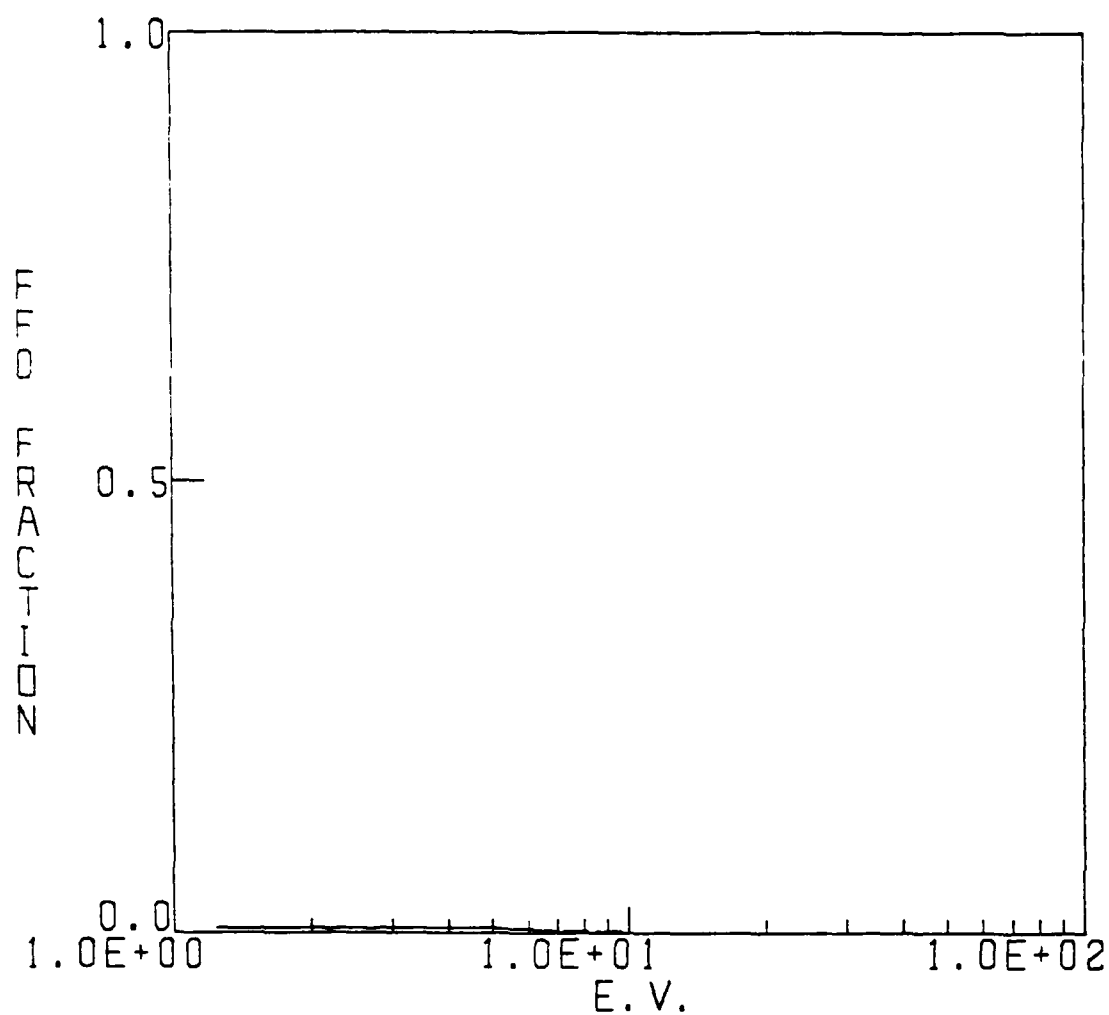


Fig. 3g — Same as Figure 1g. $T_e = 3.0$ e.v.

TE=3.0 NE=3.677E19

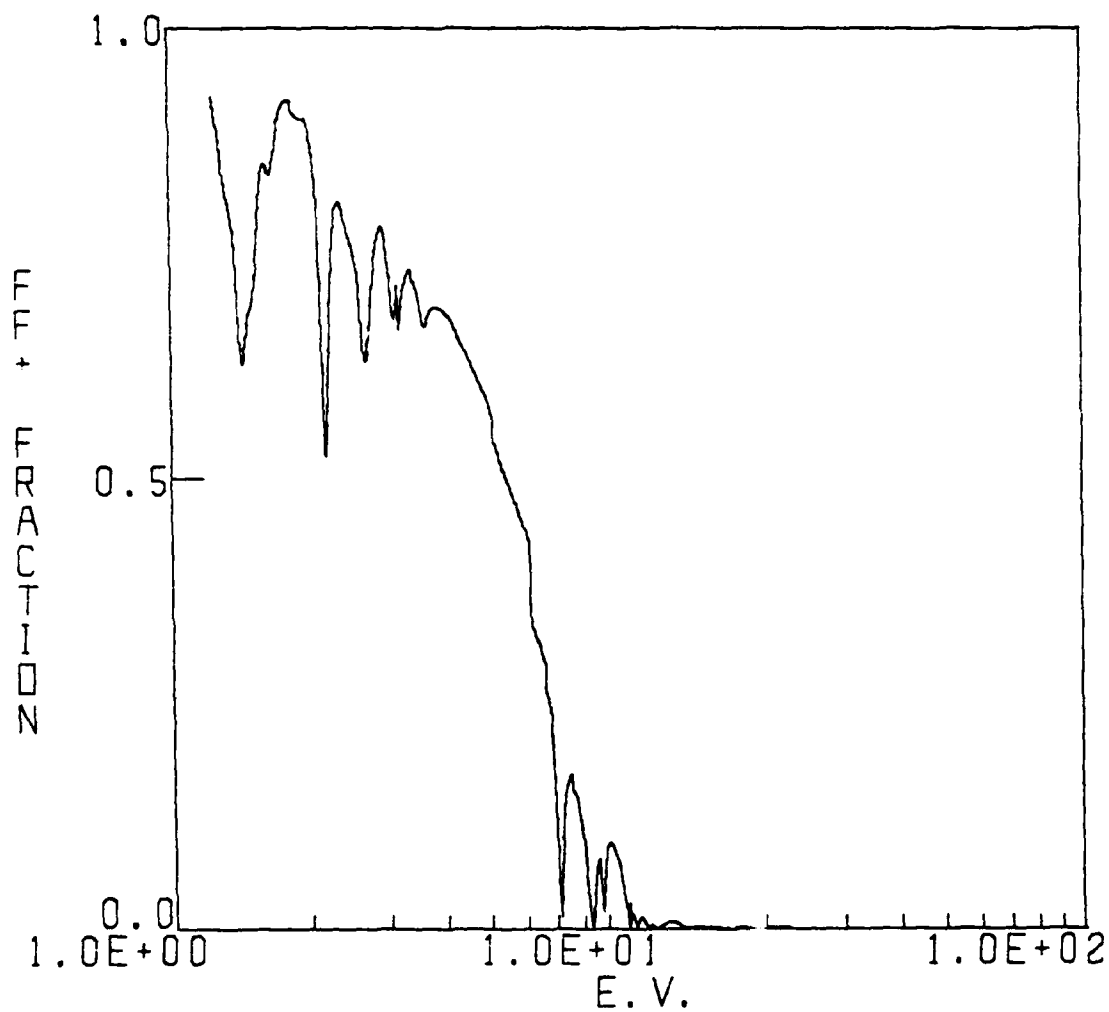


Fig. 3h — Same as Figure 1h. $T_e = 3.0$ e.v.

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